



Practical Activity

Comparing the energy of two fuels

1. Apparatus

2. Method

3. Results

Fuel	Temperature ($^{\circ}$ C) of water before.	Temperature ($^{\circ}$ C) of water after heating.

4. Conclusion

1. What a fuel is?
2. Which Fuel released most energy? Explain in a clear way all your conclusions.

PGCE secondary in science



ST. MARY'S COLLEGE
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**Unit H2 – Energy from
chemical reactions**